

# A $H^+/Ca^{2+}$ Antiporter in Membranes of Microsomal Vesicles from Maize Coleoptiles, a Secondary Energized $Ca^{2+}$ Pump

A. Hager and P. Hermsdorf

Institut für Biologie I, Universität Tübingen, Auf der Morgenstelle 1, D-7400 Tübingen 1

Z. Naturforsch. **36 c**, 1009–1012 (1981); received September 2, 1981

$Ca^{2+}$  Pump,  $H^+$ -ATPase, Membrane Vesicles, Coleoptile, *Zea mays*

Microsomal vesicles prepared from the ER and the Golgi apparatus of maize coleoptiles possess an ATP-fueled,  $Cl^-$ -dependent proton pump which can cause an intravesicular acidification of the vesicles (A. Hager and M. Helmle, Z. Naturforsch. **36 c**, 997 (1981); A. Hager *et al.*, Z. Naturforsch. **35 c**, 783–793 (1980)). This acidification is very specifically inhibited by  $Ca^{2+}$ . The inhibition is already distinct at concentrations of  $30 \mu M$ . While 10-fold higher concentrations of  $La^{3+}$  produce similar effects,  $Mg^{2+}$  remains ineffective even at very high concentrations.

If  $Ca^{2+}$  is added after acidification of the vesicles, a rapid  $H^+$ -efflux proportional to the amount of  $Ca^{2+}$  added characterizes the first reaction phase. In the second reaction phase acidification within the vesicles commences anew, however, at a reduced rate.

If  $Ca^{2+}$  is added to vesicles whose proton pump has been inhibited by DCCD, the first reaction phase remains unchanged, while the acidification in the second reaction phase does not set in, and only a leak out of protons is observed.

These results give support for a  $H^+/Ca^{2+}$ -antiporter mechanism which can function as a secondary energized  $Ca^{2+}$ -pump and regulate the  $Ca^{2+}$ -concentration in the cytoplasm.

## Introduction

$Ca^{2+}$  participates in the regulation of a variety of cellular processes [3, 4]. In plants, it has been suggested that  $Ca^{2+}$  is involved in growth-related mechanisms, in seismonastic movement of leaves, in phototaxis, in microtubuli disaggregation, in phytochrome-mediated chloroplast movement, in regulation of enzyme activities (as quoted by [5]) and in the induction of polarity [6]. The signal for the various  $Ca^{2+}$ -regulated systems is usually a transient increase in free cytoplasmic  $Ca^{2+}$  concentrations from resting levels of  $10^{-6}$ – $10^{-8}$  M to higher levels [7]. In many of the  $Ca^{2+}$ -mediated processes the  $Ca^{2+}$ -dependent regulator protein, calmodulin, is the coupling link [8]. Because a low cytoplasmic  $Ca^{2+}$  activity has to be maintained against high  $Ca^{2+}$  concentrations outside the cell and inside various cell organelles, active  $Ca^{2+}$  transport systems are necessary on cell membranes. Such  $Ca^{2+}$  pumps have been demonstrated for mitochondria [5, 9, 10] and for microsomal fractions [5, 10–12]. In general, three types of  $Ca^{2+}$  pumps have been postulated so far:

(1)  $Mg^{2+}$  +  $Ca^{2+}$ -activated ATPases located in the surface membranes and endoplasmic reticulum of various animal cells [13] and in plant membrane vesicles [5, 10, 11, 14].

(2) Mitochondrial  $Ca^{2+}$  pumps utilizing ATP or oxidizable substrates as energy source [4].

(3)  $Na^+/Ca^{2+}$  and  $H^+/Ca^{2+}$  exchange systems using the electrochemical gradient of  $Na^+$  or  $H^+$  for  $Ca^{2+}$  translocation [4, 15–17].

In the preceding paper [1] it was shown that the cytoplasm of maize coleoptiles contains vesicles which are derived from the endoplasmic reticulum or the Golgi apparatus, and which possibly can fuse with the plasmalemma or the tonoplast. Energy-dependent proton pumps which cause an intravesicular acidification are located in the membranes of such “microsomal” vesicles. The amount of acidification is strongly dependent on the  $Ca^{2+}$  concentration of the medium. In the present paper it is demonstrated that this  $Ca^{2+}$  effect is an expression of a  $H^+/Ca^{2+}$  antiporter located within the vesicle membrane. This antiporter may function as a secondary energized  $Ca^{2+}$  pump to remove  $Ca^{2+}$  from the cytoplasm.

## Materials and Methods

The preparation of the vesicles and the determination of pH within the vesicles have been described in a preceding paper [1].

**Abbreviations:** CCCP, carbonylcyanide-*m*-chlorophenylhydrazine; DCCD, N,N'-dicyclohexylcarbodiimide; EGTA, ethylene glycol-bis-( $\beta$ -aminoethyl ether) N,N'-tetraacetic acid; ER, endoplasmic reticulum.

Reprint requests to Prof. Dr. A. Hager.

0341-0382/81/1100-1009 \$ 01.00/0



Dieses Werk wurde im Jahr 2013 vom Verlag Zeitschrift für Naturforschung in Zusammenarbeit mit der Max-Planck-Gesellschaft zur Förderung der Wissenschaften e.V. digitalisiert und unter folgender Lizenz veröffentlicht: Creative Commons Namensnennung-Keine Bearbeitung 3.0 Deutschland Lizenz.

Zum 01.01.2015 ist eine Anpassung der Lizenzbedingungen (Entfall der Creative Commons Lizenzbedingung „Keine Bearbeitung“) beabsichtigt, um eine Nachnutzung auch im Rahmen zukünftiger wissenschaftlicher Nutzungsformen zu ermöglichen.

This work has been digitalized and published in 2013 by Verlag Zeitschrift für Naturforschung in cooperation with the Max Planck Society for the Advancement of Science under a Creative Commons Attribution-NoDerivs 3.0 Germany License.

On 01.01.2015 it is planned to change the License Conditions (the removal of the Creative Commons License condition “no derivative works”). This is to allow reuse in the area of future scientific usage.

## Results and Discussion

Evidence for ATP-fueled,  $Cl^-$ -dependent proton pumps in vesicles prepared from the cytoplasm of maize coleoptiles [1] has raised the question concerning their importance and regulation. In the following investigation it is demonstrated that the intravesicular  $H^+$  accumulation is very sensitively regulated by the well known secondary messenger,  $Ca^{2+}$ -ion.

Since very low concentrations of  $Ca^{2+}$  are already able to induce manifold physiological reactions, a prerequisite of any study investigating the effect of  $Ca^{2+}$  on membrane fractions is the removal of  $Ca^{2+}$  prior to experimentation. This was to some extent achieved by adding 10 mM EGTA to the homogeni-

zation medium [III] and repeated washing of the vesicle fraction (centrifugation of the vesicle fraction at  $143\,000 \times g$  and resuspending in medium IV).

Fig. 1 shows that the intravesicular ATP-dependent acidification is already inhibited by low  $Ca^{2+}$  concentrations. After 10 min  $50 \mu M$   $Ca^{2+}$  inhibit about 20%,  $500 \mu M$   $Ca^{2+}$  about 52%.

Plausible mechanisms underlying this  $Ca^{2+}$ -induced inhibition are 1. direct inhibition of the proton pump, 2. increased permeability to  $H^+$ , or 3. augmented  $H^+$ -export due to the presence of a  $H^+/Ca^{2+}$  antiporter [15–17]. The course of the curve of ATP-dependent acidification after a later  $Ca^{2+}$  addition supports the third possibility: immediately after the addition of  $Ca^{2+}$ , the energy dependent acidification is stopped (Fig. 1), and a

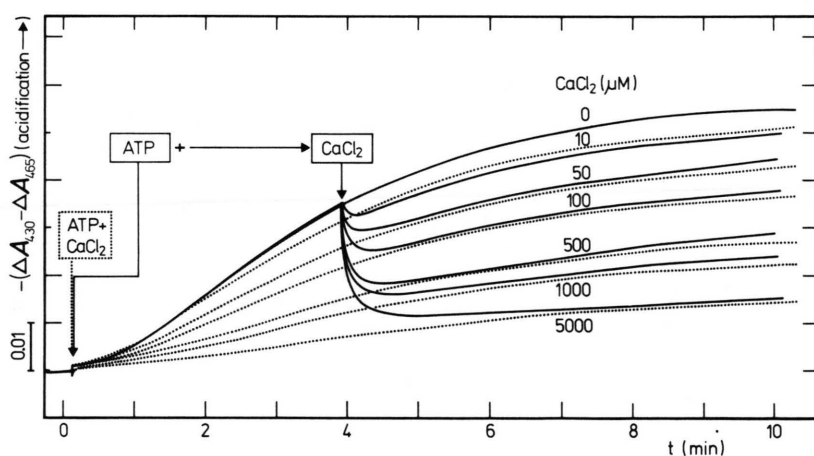


Fig. 1. ATP dependent acidification of microsomal vesicles and inhibition by initial and later additions of  $Ca^{2+}$ . Conditions as described by Hager and Helmle ([1], Fig. 1) with the exception that remaining EGTA was removed by repeated centrifugation of the vesicles ( $143\,000 \times g$ , 30 min) and resuspension in medium IV.

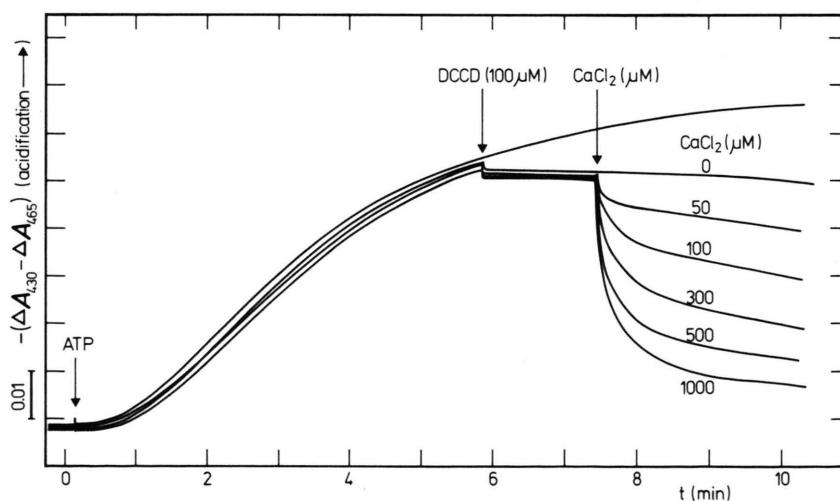
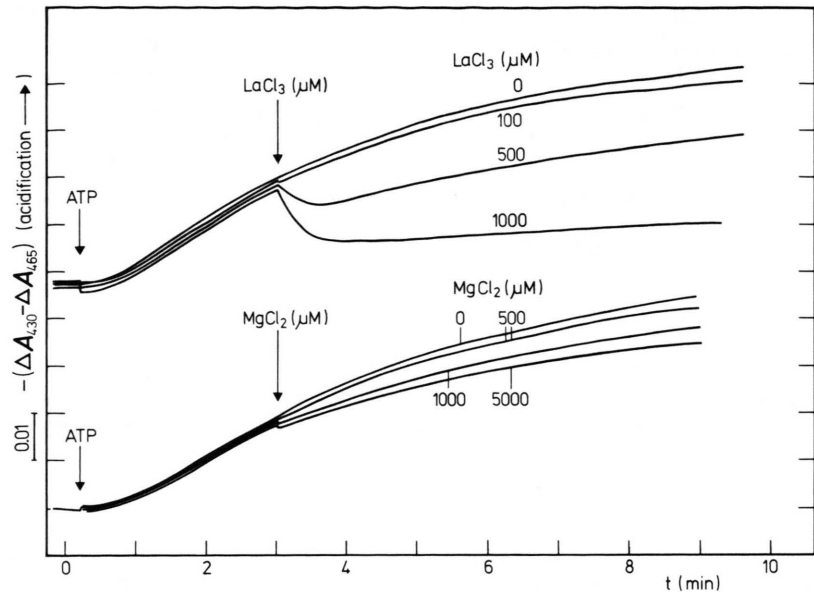


Fig. 2. ATP-dependent acidification within the vesicles; inhibition of the proton pump by DCCD, and  $H^+$ -export from within the vesicles after addition of  $Ca^{2+}$ , under the conditions described in Fig. 1.

Fig. 3. ATP-dependent acidification within membrane vesicles; release of  $H^+$ -ions induced by  $La^{3+}$  in comparison to  $Ca^{2+}$  (Fig. 1 and 2) lower,  $Mg^{2+}$  remains ineffective even at very high concentrations.



fast alkalinization is observed, that is to say a release of  $H^+$  from within the vesicle into the buffered environmental solution. This export is strictly proportional to the added amount of  $Ca^{2+}$ . After this first reaction phase lasting about 1–2 min, the energy dependent acidification in the vesicles is restored, but at a reduced speed. The lower rate of acidification in the vesicles determined now corresponds approximately to the rate determined where ATP and  $Ca^{2+}$  were simultaneously added at the beginning of the experiment (Fig. 1).

The sudden efflux of  $H^+$  from inside the vesicles after the addition of  $Ca^{2+}$  can best be ascribed to a  $H^+/Ca^{2+}$  antiporter located in the membrane of the vesicle, which renders a  $Ca^{2+}$ -influx into the vesicles in exchange for  $H^+$  until a balance in concentration is reached or a new thermodynamic equilibrium established. Then an ATP-dependent  $H^+$ -accumulation takes place anew, however, at a lower speed, since  $H^+$  can still be further exchanged by  $Ca^{2+}$ . However, since  $Ca^{2+}$  now has to be transported uphill against the diffusion gradient, the antiporter prevents a complete efflux of  $H^+$  and causes an accumulation of  $H^+$  at a lower rate.

If the  $H^+$ -pumps on the membrane are inhibited by DCCD (100  $\mu M$ ), an addition of  $Ca^{2+}$  again brings about an incipient, fast export of  $H^+$ , however, a subsequent re-acidification can not be observed, but only a slow, rather unspecific leak out of

$H^+$  from the vesicles. This slow  $H^+$ -efflux is the same for all concentrations of  $Ca^{2+}$  added.

While  $La^{3+}$  which in various tissues readily displaces  $Ca^{2+}$  from its binding sites [18] can first achieve a similar effect as  $Ca^{2+}$ -ions at substantially higher concentrations,  $Mg^{2+}$  remains completely ineffective (Fig. 3). This speaks in favor of a  $H^+/Ca^{2+}$ -antiporter with a very specific binding site for  $Ca^{2+}$ .

The  $H^+/Ca^{2+}$ -antiporter mechanism can be regarded as a secondary energized  $Ca^{2+}$  pump. Such has been detected in the plasmalemma of *Neurospora* [15–17].

Furthermore, a similar mechanism seems to exist in sarcoplasmic reticulum vesicles, since  $H^+$ -efflux is fairly tightly coupled with  $Ca^{2+}$  uptake utilizing a counter transport mechanism [19–21].

The function *in vivo* of the  $H^+/Ca^{2+}$ -antiporter described here could be to keep the cytoplasm free of  $Ca^{2+}$  ions or to adjust the  $Ca^{2+}$  level according to physiological conditions. Within the vesicles themselves then, an augmented  $H^+$ -accumulation could first take place only after a decrease of the  $Ca^{2+}$  concentration in the cytoplasm.

The demonstrated  $Ca^{2+}$ -pumping mechanism does not seem to be identical with the  $Ca^{2+}$ -pump described by Gross and Marmé (1979) [11] on microsomal vesicles from zucchini. The latter  $Ca^{2+}$  uptake system cannot be inhibited by CCCP and it shows

no such high specificity for ATP. It is possible that these two Ca<sup>2+</sup> transport systems are located on different membrane vesicles and that they have different tasks in regulating cell metabolism.

#### Acknowledgements

We thank Miss Karin Putz for skillfull technical assistance and Dr. J. Gross for critical and stimulating discussions.

- [1] A. Hager and M. Helmle, *Z. Naturforsch.* **36 c**, 997–1008 (1981).
- [2] A. Hager, R. Frenzel, and D. Laible, *Z. Naturforsch.* **35 c**, 783–793 (1980).
- [3] A. Martonosi, *Fed. Proc.* **39**, 2401–2402 (1980).
- [4] E. Carafoli and M. Crompton, *Current topics in membranes and transport* **Vol. 10**, New York, Academic, 151–216 (1978).
- [5] P. Dieter and D. Marmé, *Planta* **150**, 1–8 (1980).
- [6] M. H. Weisenseel, in: *Encyclopedia of Plant Physiol. New Series* **Vol. 7**, 485–505, Springer, Berlin-Heidelberg 1979.
- [7] F. L. Bygrave, *TIBS* **3**, 175–178 (1978).
- [8] H. Charbonneau and M. J. Cormier, *Biochem. Biophys. Res. Commun.* **90**, 1039–1047 (1979).
- [9] R. H. Wilson and R. J. Graesser, in: *Encyclopedia of Plant Physiol. New Series* **Vol. 3**, 377–397, Springer, Berlin, Heidelberg, New York 1976.
- [10] P. Dieter and D. Marmé, *Proc. Nat. Acad. Sci. USA* **77**, 7311–7314 (1980).
- [11] J. Gross and D. Marmé, *Proc. Nat. Acad. Sci. USA* **75**, 1232–1236 (1978).
- [12] J. Gross, in: *Plant Membranes*, D. Marmé (ed.) Elsevier, North Holland 1981, in press.
- [13] L. de Meis and A. L. Vianna, *Ann. Rev. Biochem.* **48**, 275–292 (1979).
- [14] P. Dieter and D. Marmé, *FEBS Letters* **125**, 245–248 (1981).
- [15] P. Stroobant, J. B. Dame, and G. A. Scarborough, *Fed. Proc.* **39**, 2437–2441 (1980).
- [16] P. Stroobant and G. A. Scarborough, *Proc. Nat. Acad. Sci. USA* **76**, 3102–3106 (1979).
- [17] G. A. Scarborough, *Biochemistry* **19**, 2925–2931 (1980).
- [18] R. B. Martin and F. S. Richardson, *Quart. Rev. Biophys.* **12**, 181–209 (1979).
- [19] T. Ueno and T. Sekine, *J. Biochem.* **89**, 1239–1246 (1981).
- [20] T. Ueno and T. Sekine, *J. Biochem.* **89**, 1247–1252 (1981).